A study of the effectiveness and energy efficiency of ultrasonic emulsification

Wu Li^{a,b}, Thomas S.H. Leong^{a,b,c}, Muthupandian Ashokkumar^{a,b}, Gregory J.O. Martin^{a,c,*}

Received 00th January 20xx,
Accepted 00th January 20xx

DOI: 10.1039/x0xx00000x

www.rsc.org/

Three essential experimental parameters in the ultrasonic emulsification process, namely sonication time, acoustic amplitude and processing volume, were individually investigated, theoretically and experimentally, and correlated to the emulsion droplet sizes produced. The results showed that with a decrease in droplet size, two kinetic regions can be separately correlated prior to reaching a steady state droplet size: a fast size reduction region and a steady state transition region. In the fast size reduction region, the power input and sonication time could be correlated to the volume-mean diameter by a power-law relationship, with separate power-law indices of -1.4 and -1.1, respectively. A proportional relationship was found between droplet size and processing volume. The effectiveness and energy efficiency of droplet size reduction was compared between ultrasound and high-pressure homogenisation (HPH) based on both the effective power delivered to the emulsion and the total electric power consumed. Sonication could produce emulsions across a broad range of sizes, while high pressure homogenisation was able to produce emulsions at the smaller end of the range. For ultrasonication, the energy efficiency was higher at increased power inputs due to more effective droplet breakage at high ultrasound intensities. For HPH the consumed energy efficiency was improved by operating at higher pressures for fewer passes. At the laboratory scale, the ultrasound system required less electrical power than HPH to produce an emulsion of comparable droplet size. The energy efficiency of HPH is greatly improved at large scale, and may also be higher for larger scale ultrasonic reactors.

1 Introduction

In its simplest form, an emulsion is a mixture of two immiscible liquids usually consisting of an organic “oil” phase (O) and a water phase (W), in which one liquid (the dispersed phase) is in the form of microscopic droplets dispersed in the bulk (continuous) phase. Two common types of emulsions are the oil-in-water (O/W) and water-in-oil (W/O) emulsions. To stabilize the two immiscible liquids, emulsifier is usually required to prevent spontaneous separation back into individual phases. The choice of the emulsifier is an important consideration in the emulsion formation and its long-term stability^{1–3}. Emulsions have wide applicability in a range of fields including paints, foods, cosmetics and drugs^{4–7}. For many applications, the emulsion droplet size (EDS) of the disperse phase is a key factor that governs the stability, appearance and functionality of the produced emulsion. With a decrease in EDS from a size range of 0.5–100 μm (i.e., macroemulsion) to a size range of 20–200 nm (i.e., nanoemulsion), a higher kinetic stability can be achieved⁶.

Additionally, the size reduction process can lower the turbidity of the emulsion, potentially creating high functionality without affecting the appearance of the final product⁸. The selection of appropriate equipment and processing parameters is important to enable the formation of nano-sized emulsions required for optimally designed products. The emulsification technique is critical for achieving a nanoemulsion size range and there are many techniques available. “Low energy” emulsification methods generally involve two approaches: the transitional inversion and the spontaneous inversion, where the hydrophilic-lipophilic balance (HLB) of the former systems will be drastically altered by changing the physical or chemical parameters, such as temperature and electrolyte concentration; the later approach can be achieved by varying the dispersed phase volume fraction^{9–11}. Several limitations can be found in the low energy methods. For instance, the large amount and particular type/combination of surfactants are not generally applicable to large-scale industrial production¹². The industrial production of sub-micron emulsions is primarily based on the application of high shear to disrupt and distribute a disperse phase into small droplets throughout the continuous phase. High-pressure homogenizers (HPH) are one of the most effective methods by which nanoemulsions can be formed. In these devices, a high pressure is used to drive fluid flow across very narrow valve gaps, creating extreme turbulence, cavitation, high

^a ARC-ITRP Dairy Innovation Hub, The University of Melbourne, Parkville, Victoria 3010, Australia.

^b School of Chemistry, The University of Melbourne, Parkville, Victoria 3010, Australia.

^c Department of Chemical Engineering, The University of Melbourne, Parkville, Victoria 3010, Australia.

shear and inertial forces¹³. With homogenizing pressures of up to 2500 bar achievable, it is possible to create emulsions with a mean droplet diameter of less than 0.2 μm ¹⁴. HPHs are widely used in industry (e.g., dairy processing), however they require regular maintenance due to the moving parts and the susceptibility of the values to damage. Ultrasonic systems that generate high local power densities offer an alternative to HPH systems. The physical shear and turbulence generated in ultrasonic systems originates from the phenomenon of acoustic cavitation. Due to the pressure fluctuation of the sound wave throughout the aqueous medium, gas pockets in water act as nuclei, and microbubbles can form, grow and collapse within microseconds, causing implosions at a micro-scale, providing intense localised shear forces. The cumulative effect of these microscale-implosions throughout the medium gives rise to ultrasonic emulsification^{5, 15–17}. The high intensity of acoustic cavitation in low-frequency systems can generate several high-impact physical effects such as micro-streaming, micro-jetting and shockwaves^{18–20}. As a result, high turbulence and velocity gradients can be generated over very small length scales, leading to the production of nanoemulsion^{8, 21, 22}. For ultrasonic emulsification, a well-accepted mechanistic theory has been established by Li and Fogler, which involved a two-step mechanism starting from two separate phases. It was concluded that the higher the ultrasonic frequency, the more difficult it is to reach the critical amplitude of acoustic vibration required to initiate the mixing of the two separate phases²³. Further, with an increase in frequency, the radius of the cavitation bubbles decreases, resulting in less violent collapse of the bubbles, such that the physical effects (e.g. shockwaves) gradually weaken while the chemical effects (i.e. radical formation) strengthen. Therefore, low ultrasound frequencies, typically 20 kHz, are commonly used for emulsification purposes. Regardless of the technique, emulsification requires considerable energy, and it is important to understand how to maximise energy efficiency. Leong et al. have examined the correlation between the energy density (sonication energy per unit volume) and the Z-average droplet size as a function of the sonication time⁸. Across various sonication setups (i.e. batch and continuous systems with different ultrasonic transducers and container designs) operated at ambient pressure, the correlation of droplet size and energy density showed a power law index of -0.37⁸. Abismail et al. investigated the relation between droplet size and sonication power density, ϵ (power input per unit volume)²⁴, without investigating the effect of changing the processing volume. Similar measurements were reported by Gupta et al.^{21, 22}, where three different ultrasound amplitudes were employed using the same emulsification container. These studies have demonstrated that ultrasonic systems are capable of producing emulsions spanning across a size range of submicron to nanometer scales. However, to date the kinetics of droplet size reduction has not been well studied in the context of ultrasonic systems

and the maximisation of energy efficiency. A comprehensive correlation between droplet size and energy density still needs to be established based on the combination of individual parameters (i.e. power input, sonication time and processing volume).

To the best of our knowledge, there is no systematic study available in the literature that focused on correlating the EDS to various ultrasonic experimental parameters, both theoretically and experimentally. In order to fill this knowledge gap, this work aims to understand, from both theoretical and experimental viewpoints, the effect of selected experimental parameters of ultrasonic emulsification on the kinetics of EDS reduction, including sonication amplitude, residence time and processing volume using a select surfactant/oil system. Instead of using power density or energy density, the three parameters will firstly be examined independently in order to explore their individual effects in more detail. The results will be interpreted on the basis of the physics of emulsification as well as providing an empirical relationship. The energy efficiency of US emulsification will also be benchmarked against HPH.

Theoretical background

The theory around emulsification in turbulent flow was established half a century ago and a comprehensive understanding has been established on the prediction of macroemulsion droplet size as a function of power density^{25–29}. Two regimes of emulsification have been distinguished: the turbulent inertial and the turbulent viscous regime²⁹. The difference between the two regimes is the relative size of the droplets, d , to the smallest eddies, λ_0 . In the turbulent inertial regime, $d > \lambda_0$, whereas in the turbulent viscous regime, $d < \lambda_0$. The size of the smallest eddies in the turbulent flow, λ_0 , which is also referred to as the “Kolmogorov scale”, can be defined according to the following empirical relationship that reflects the hydrodynamic conditions during emulsification:

$$\lambda_0 \approx \epsilon^{-1/4} \eta_c^{-3/4} \rho_c^{-3/4}, \quad (1)$$

where η_c is the viscosity (m^2s^{-1}) and ρ_c is the mass density of the continuous phase (kg m^{-3}), while ϵ is the average power dissipated per unit mass of the fluid (W kg^{-1}).

In the turbulent inertial regime, there is a critical droplet size at which there is a balance between droplet breakup and recoalescence. The critical stable drop diameter, d_{crit} is expressed as follows²⁹:

$$d_{crit} = C \epsilon^{-2/5} \sigma^{3/5} \rho_c^{-3/5}, \quad (2)$$

where d_{crit} is for the critical droplet size at a low dispersed phase viscosity ($\eta_d < 10 \text{ mPa s}$)⁴, C is a constant, ϵ is the power density (i.e., the average power dissipated per unit mass), σ is interfacial tension and ρ_c is the density of continuous phase. In recent studies, the critical diameter has often been represented using different diameter presentations (e.g., volume mean, Sauter mean and Z-average diameter), with the correlation generally holding but with altered values for the constant^{8, 21, 30}.

Equation (2) gives a prediction of droplet size based on power density as $d \sim \epsilon^{-0.4}$ for the formation of macroemulsions (0.1 – 5 μm) and has been widely applied in the studies of different systems such as ultrasound (US), Ultra-Turrax (UT) and HPH systems to determine the difference in energy efficiency³⁰. However, due to the lack of time dependence, equation (1) is only suitable to describe the critical diameter in the steady state where an equilibrium between droplet breakage and coalescence has been reached. In other words, the impact of residence time on droplets' disruption kinetics cannot be elucidated. During practical processes, although the critical droplet size is essential to characterize the theoretical limit of different techniques, the kinetics of droplet breakup is also of significant importance to predicting the results. The kinetics of size reduction depend on the residence time within the actual dispersing zone, which for high energy systems, is where an extreme level of energy is focused in a small area³¹. Typically, in US systems, uneven power density distribution in the dispersing zone likely causes a wide EDS distribution. There is, therefore, a strong dependence of droplet size on the residence time that needs to be taken into consideration.

Another commonly reported quantity is the energy density (E_v), which is the energy input per unit volume, per unit time^{8, 31}:

$$E_v = P\tau/V = P/v, \quad (3)$$

as P is the power input (W), τ is the residence time (s), V is the processing volume (mL) and v is the flow rate (mL s⁻¹). The non-steady state average droplet diameter has also been described by the following power-law relationship^{8, 14, 17, 31}:

$$d(t) = C \epsilon^{b1} \cdot \tau^{b2}, \quad (4)$$

where $b1$ and $b2$ are the power-law index of power density and residence time. For ultrasound systems, it has been suggested that the power density and residence time may exert different effects on the kinetics of droplet breakup³¹. Yet, the difference between the two parameters has not been clearly demonstrated in previous work, and correlations need to be established for practical proposes. Therefore, it is important to investigate the effect of individual parameters in more detail. The three parameters that can be separately investigated from the equations mentioned above are the power input P , sonication time τ and processing volume V . Here we look to consider the kinetics of emulsification as a function of power and volume, which can be represented as follows:

$$d(t) = C V^a P^{-b} \tau^{-c}, \quad (5)$$

Relevant studies have been conducted on only either the power input or sonication time. For instance, Walstra³⁰ has compared the volume mean diameter of emulsion droplets as a function of energy density amongst UT (batch), US (continuous) and HPH (continuous) systems by only varying power intensity ϵ with dilute paraffin O/W emulsions. However, since the stirring time for UT system was fixed at 2 min and both US and HPH were processed at a constant flow rate, the correlations could only be presented based

on power density, ϵ , consistent with equation (2). As such, it was not possible to give consideration to the kinetics, or a correlation with the residence time. From a practical perspective, it is important to examine the impact of different parameters and to establish correlations that can help improve the energy efficiency of emulsification processes.

Here we look to develop an understanding of the impact of power input, sonication time and processing volume on the kinetics of emulsification. The potential difference amongst the parameters will be explored by examining the emulsion droplet breakup over a wide range of conditions. Correlations will be determined based on the power-law fit to provide empirical relationships that allow comparison of different types of mechanical emulsification equipment from energy consumption and emulsification efficiency perspectives for potential scaling-up purposes in practical applications.

Materials and methods

Materials

Sodium dodecyl sulfate (SDS) (Sigma Aldrich) and sunflower oil (Woolworths, Australia) were used as the surfactant and oil phase, respectively, without any further modification. Milli-Q water was used in all experiments as the continuous phase with a typical resistivity of 18.2 M Ω ·cm at 25°C.

Methods

Impact of relevant parameters of ultrasonic emulsification

The sonication process was conducted using a Branson Digital Sonifier (Branson, Connecticut, Model No. 450, nominal power 400 W) connecting to a transducer (No. 102C) with a 12 mm diameter horn tip at a frequency of 20 kHz. The batch and continuous flow (CF) ultrasound systems were performed using the experimental setup shown in Fig. 1.

One set of experiments was designed to investigate the influence of processing volume, but fixed interfacial contact area between the horn and the fluid. For this, three conical flasks of different volume but with a similar internal neck diameter around 2 cm were used. The processing volumes were adjusted according to the neck position to ensure a similar oil surface area contacted with the horn tip.

Fig. 1 Schematic illustration of the ultrasound emulsification batch (A) and continuous flow (B) systems.

The effect of three parameters: sonication amplitude, time and processing volume were investigated as follows. Sonication amplitude was varied from 10% to 50% of the total power (400 W) (details of the calorimetric power characterization are shown below). The sonication time was varied from 30 s to 10 min for each amplitude applied. Three different processing volumes, 60, 110 and 180 mL, were examined using three conical flasks of different sizes. The three conical flasks had a similar neck width of ~2 cm. The emulsions were prepared with fixed SDS concentration of 5 mM in Milli-Q water and 1 wt% of canola oil to reduce impact from dispersed phase viscosity, close packing limits. For surfactants in an emulsion system, there exists a competition between the oil/water and air/water interfaces, which will exert an important effect on the size of O/W emulsion droplets that are formed. SDS surfactant is well-known to foam when subject to strong agitation. During ultrasonication, acoustic cavitation also creates air bubbles, causing some of the SDS molecules to move to the air/water interface. This can result in the creation of SDS-stabilised foams, which will cushion the shockwaves generated by sonication. The SDS concentration, sonication power input and the position of the sonicator probe, all need to be carefully adjusted to avoid excessive foaming. Some preliminary experiments were conducted whereby the parameters mentioned were adjusted to avoid onset of excessive foaming during the sonication period (not shown). The occurrence of foaming resulted in poorer emulsification (evident from a damping of the sonication noise) and the formation of larger than expected droplets. Further, some of the oil droplets became immobilised in the foaming layer at the top of the aqueous phase, limiting their ability to be processed by the applied ultrasound. There was a threshold sonication power which, once exceeded, lead to excessive foaming and inefficient emulsification. Similarly, this was also the case with the SDS concentration. Based on these findings, an 'optimal' SDS concentration of 5 mM and ultrasonic power of 50% amplitude (200 W) was found to enable emulsification without formation of excessive foaming, even for prolonged durations of ultrasonication. The solution temperature was controlled by the circulation of cooling water at $22.5 \pm 2^\circ\text{C}$. The conical flasks were placed

in a glass cell with circulating water at sufficient level for submerging the entire emulsion.

Droplet size reduction and energy consumption comparison

The emulsions for the comparison of three systems were prepared as follows. US batch emulsions were prepared by employing the same experimental setup as mentioned above. US CF emulsions were prepared by using Ultraturrax (IKA-Labortechnik) as the pre-emulsification step at the lowest speed (6500 rpm) for 120 s to avoid excessive foaming, followed by applying sonication at 50% amplitude (160W) with the residence time varied from 0.45 s to 9 s controlled by a peristaltic pump (Baoding Longer Precision Pump Co., Ltd. Equipped with YZ1515x pump head and 16" Tubing Type). The preparation of HPH emulsions involved a pre-emulsification using UT at the same setup as US pre-emulsification step. The coarse emulsions were then immediately transferred to the HPH system GEA PandaPLUS 1000 (GEA Niro Soavi, Parma, Italy). A single stage/pass homogenisation was performed with an increase of pressure on the pressure valve from 20 MPa to 120 MPa with a constant flow rate of 2.77 mL/s. To extend the range of energy density, some emulsions were processed under multiple passes using the sample collected after the previous pass at the same pressure condition.

Emulsion droplet size and size distribution

The droplet size and size distribution of emulsions were measured using a Malvern Mastersizer 3000 Particle Size Analyzer. A refractive index of 1.462 and absorption of 0.001 was used for the measurements. Each measurement was repeated three times. The mean droplet diameter evaluated in this work is expressed as volume-mean diameter $d[4,3]$. The choice of using volume mean diameter was decided based on the size distribution range of samples. The expression of $d[4,3]$ can be represented as follows:

$$d[4,3] = \sum n_i d_i^4 / \sum n_i d_i^3 \quad (6)$$

Electrical power consumption

For the sonicator and the Ultra-Turrax unit, a single-phase energy cost meter was used to determine the electric power drawn. An in-built three-phase energy cost meter was used to measure the power drawn from the high-pressure homogeniser. Stable readings were recorded in triplicate after the fluctuations that may occur at the beginning of operations.

Calorimetry calibration of the ultrasound system

The calorimetric power delivered by the horn transducer was determined using 100 mL Milli-Q water in a glass cell. A 1 min long sonication step at 10%, 20%, 30%, 40% or 50% amplitude was conducted, with the temperature pre- and post-sonication recorded. Using these temperatures, the calorimetric power (P_{calo}) was calculated using the following equation¹⁷:

$$P_{\text{calo}} = cm\Delta T/t, \quad (7)$$

where c is the heat capacity of water, m is the mass of water, ΔT the temperature increase caused by sonication, and t sonication time.

Statistical analysis

The correlations amongst droplets size and ultrasonic parameters were established by conducting linear or exponential regression depending on different parameters. The coefficient of determination R^2 was used to evaluate the performance of regressions. Unless otherwise stated, all the data were collected based on triplicated experiments with the standard deviation shown in error bars. In addition, all analytical measurements conducted in this work were performed in triplicates.

Results and discussion

Impact of ultrasound power input on the kinetics of emulsion droplet size reduction

The volume-mean diameter of the droplets as a function of sonication time and power intensity is presented in Fig. 2. As expected, applying ultrasound at higher power input (for a given volume) produced smaller emulsion droplets more rapidly than at lower power. For all power inputs tested, the average droplet size was approximately inversely proportional to sonication time, reflecting the increasing difficulty of breaking apart progressively smaller droplets. By fitting the experimental data before reaching steady state transition using a power law fit, power-law indices of ~ 1.1 were determined, with high $R^2 = 0.99$, regardless of the power amplitude used. This indicates that the proportionality between sonication time and droplet size is similar at various power levels (i.e., for a given emulsion and sonication power, halving the average droplet diameter will require processing for just under twice the amount of time). As many studies have demonstrated, a steady state (i.e., where the droplet size does not decrease with further sonication) is eventually reached as the rate of shear-induced droplet breakup decreases to match the rate of droplet recoalescence^{8, 21, 22, 24, 29, 32}. After prolonged ultrasonication at a power of 120 W and above (see triangular data points in Fig. 2) the kinetics deviate from the established power law trend lines (dashed lines) as the droplet size starts to approach a steady state. To approximate the transition in kinetics from the initial power law to steady state, new power law trend lines can be extracted from the existing data, which share a power-law index value around -0.6. Although steady state was not fully reached in even the highest power input, the reduce rate of size decrease is evident in the transition region. This reflects the increasing difficulty of breaking smaller droplets³³. In a previous study by Cucheval and Chow³⁴, it was shown that a steady state diameter was reached after shorter (3 min) durations of sonication at a similar power input used in this current study. It should be noted however that the droplet sizes reached by Cucheval and Chow, were approximately

0.7 μm for all power inputs investigated. Compared to the results in this study and others^{8, 21, 22}, droplet sizes can reduce to the nano-size range with further treatment time. The main difference here is the surfactant/oil system used, which suggests that the ratio of the surfactant concentration/oil volume could be a limiting factor in the size reduction ability in some systems.

Fig. 2 Volume mean droplet diameter, $d[4,3]$, as a function of sonication time for different power intensities (W) at a constant volume (60 mL). Triangular legends stand for the data points showing the transition towards steady state. Error bars represent the standard deviation of triplicate experiments.

Prior to the transition region (T-region) the power-law indices with respect to residence time were ~ 1.1 for each power intensity. To determine the power law index with respect to input power prior to the transition region, a correlation was performed between $d[4,3]/\tau^{-1.1}$ and power input (using all the pre-transition period data presented as circles in Fig. 2). As shown in Fig. 3, a power-law index for the fit was found to be around -1.5. Similarly, a correlation of droplet size and power input was plotted using the experimental data within the transition region towards steady state (i.e. the triangle symbols in Fig. 2 and 3), the power law index was only around -0.6 for both sonication time and power input. Results from such a power-law fit between droplet size and power input has not been proposed in any of the previous studies. Nonetheless, implementing the same mathematical method on the data from the work by Abismail et al. provides comparable power-law indices of around -1.6 and -0.6 within fast size reduction region and steady state transition region respectively²⁴.

449

450 Fig. 3 Correlation between $d[4,3]/\tau^{-1.1}$ and power density. Data points are the
451 average of values obtained from the corresponding data (circle legends) presented
452 in Fig. 2, where the data was normalized to the power-law index of -1.1. The error
453 bars represent the standard deviation of these values.

454 By incorporating sonication time and power input together,
455 the following equation can be devised to predict the droplet
456 size outcome at two regions:

$$d[4,3] = CP^{-1.4}\tau^{-1.1}, \quad (8)$$

$$d[4,3] = C(P\tau)^{-0.6}, \quad (9)$$

459 It has been stated that the power and residence time in
460 ultrasonic systems may provide different impacts on
461 droplet disruption³¹, and this is reflected in the relationship
462 found in equation 8. To verify equation (8) and (9), the data
463 from Fig. 2 in both regions were plotted (Fig. 4) against the
464 proposed correlation. For the data obtained at fast size
465 reduction region (circles) and steady state transition region
466 (triangles), good linear agreements (R^2 value = 0.99 and
467 0.98) were obtained.

468

469 Fig. 4 Experimental data from Fig. 2, replotted as a function of $P^{-1.4}\tau^{-1.1}$ and $P^{0.6}\tau^{0.6}$
470 (insert plot). Circles and triangles represent data obtained in fast size reduction
471 region and steady state transition region, respectively.

472 Impact of processing volume on the kinetics of emulsion 473 droplet size reduction

474 The relationship between emulsification kinetics and
475 processing volume was examined further by varying the
476 processing volume (V), and thereby the power density ϵ (W

477 mL^{-1}) at a given power intensity (40% amplitude, 160 W).
478 The calorimetric power (P_{calo}) was implemented in this
479 experiment to examine the actual amount of energy
480 dissipated for generating acoustic cavitation and shear
481 stress^{5, 17, 33}. Experiments were performed using three
482 selected volumes, which cover the processing volume range
483 of general ultrasonic emulsification processes (60, 110 and
484 180 mL, corresponding to power densities of 0.53, 0.29 and
485 0.18 $W mL^{-1}$, respectively). As expected, an increase in the
486 processing volume (decrease in the power density) reduced
487 the rate of droplet size reduction (Fig. 5A), owing to a
488 reduction in the average energy density as the processing
489 volume increases. To account for this, the same data were
490 presented as a function of energy density (Fig. 5B). A
491 correlation between $d[4,3]/\tau^{-1.1}$ and volume (insert plot in
492 Fig. 4A), was performed to determine the impact of this
493 volume change. Based on the linear trend, a proportional
494 correlation can be extracted as $d[4,3] \sim V$.

495

496

497 Fig. 5 Volume mean average droplet diameter, $d[4,3]$, as a function of sonication
498 time (A) and energy density, E_v (B) at different processing volume and a constant
499 power of 160 W. The insert plot shows the correlation between $d[4,3]/\tau^{-1.1}$ and
500 processing volume (V). Error bars represent standard deviation of triplicate
501 experiments.

502 By normalising the impact of processing volume, all the
503 experimental data presented in Fig. 5 can be regressed into
504 a linear correlation with $R^2=0.99$ (Fig. 6). The high quality
505 of the fit shows the consistency of employing the power-law
506 index of power input and sonication time proposed above

to different power densities. The result that the effectiveness and energy efficiency of droplet size reduction is independent of power density throughout the examined range, supports the possible scalability of ultrasound systems. Comparing this finding with the results above (i.e., the dependence on power intensity versus the independence on power density) indicates that it is the intensity of power in the emulsification zone of the probe that is important for achieving high energy efficiency, and that larger volumes can be processed simply by increasing the sonication time in proportion to the volume. This is consistent with the idea that it is the time the droplets spend in the emulsification zone near the horn tip that is important, and that with adequate mixing, this time will be proportional to the sonication time and inversely proportional to the processing volume. This understanding has implications for the translation of batch into continuous flow-through processes.

525

526 Fig. 6 Correlation between $d[4,3]/V$ and $P^{-1.5} \tau^{1.1}$.

527 Based on the three parameters discussed above, equation
528 10 can be presented:

$$529 \quad d[4,3] = C_1 \cdot V \cdot P^{-1.4} \tau^{-1.1}, \quad (10)$$

530 This correlation can be applied across a range of power
531 intensities and power densities in the fast size reduction
532 region mentioned before. This correlation can be reframed
533 for an energy density perspective as:

$$534 \quad d[4,3] = C_1 P^{-0.4} E_v^{-1.1}, \quad (11)$$

535 where E_v is the energy density (J mL^{-1}) defined in equation
536 (3), and C_1 is a constant. The power-law index found in this
537 work was around -1.1, which is higher in magnitude than
538 the value reported by Walstra of -0.4³⁰. The difference
539 could be attributed to the ultrasonic systems used. In this
540 study, the ultrasound was applied in a batch system
541 whereas in Walstra's study a continuous flow system was
542 employed. Due to the small dissipation zone around the
543 transducer tip, the uneven distribution of physical shear will
544 affect the average size and size distribution in both batch
545 and continuous flow systems. The effect of the uneven

546 dissipation zone can be easily overcome by increasing the
547 sonication time in a batch system. This may contribute to an
548 improved size reduction efficiency over continuous systems
549 in which it is more challenging to ensure homogeneous
550 treatment of all fluid elements in a single pass.
551 Furthermore, the restricted geometry of the oil-tip contact
552 area used in the current experiments improved the initial
553 disruption of oil droplets, which can help increase
554 performance³⁵. Additionally, the lower magnitude of
555 power-law index in Walstra's study could be because the
556 steady state transition region had been reached. In the
557 work by Leong et al. the power-law index was found to be
558 around -0.4 to -0.6 for a droplet size-energy density
559 correlation for an application of ultrasonic emulsification
560 for nanoemulsion (40-200 nm in diameter). In this instance
561 most of the EDS data were located in the transition region⁸.
562 The experiments in this study were designed such that
563 emulsification at different power intensities could be
564 directly compared at equivalent energy densities
565 (sonication time x power input/processing volume). For this
566 analysis, all the data presented in Fig. 2 is replotted on the
567 basis of energy density E_v , as shown as Fig. 7 (on the basis
568 of calorimetric power (P_{calo})). The power-law indices for the
569 data obtained prior to the transition region show good
570 agreement with equation (11) proposed above.
571 Additionally, by normalizing the constant of the power-law
572 fit between droplets size and energy density, a correlation
573 between $d[4,3]/E_v$ and calorimetric power (P_{calo}) was
574 developed with a resulting power-law index of -0.364. Of
575 practical significance, the results show that for a given
576 power density in the test range (0.13 W mL^{-1} to 0.66 W mL^{-1}),
577 it is more energy efficient to use a higher power density
578 for a shorter time before entering the steady state
579 transition (i.e. less energy will be required to produce an
580 emulsion with a given average diameter). For example, for
581 the system considered here, to produce an emulsion with a
582 volume-mean average diameter of 1.2 micron would
583 require about 55 J mL^{-1} at 50% amplitude ($P_{calo} = 40 \text{ W}$), 65
584 J mL^{-1} at 40% amplitude ($P_{calo} = 32 \text{ W}$), 75 J mL^{-1} at 30%
585 amplitude ($P_{calo} = 24 \text{ W}$), 81 J mL^{-1} at 20% amplitude ($P_{calo} =$
586 16 W), and 89 J mL^{-1} at 10% amplitude ($P_{calo} = 8 \text{ W}$). The
587 general trend of increased energy efficiency as a function of
588 increasing power intensity can be explained on the basis of
589 acoustic cavitation. The theory of cavitation bubble
590 dynamics says that with an increase in acoustic pressure, or
591 in this case power input, the number of inertial cavitation
592 bubbles (bubbles that undergo inertial collapse) and the
593 intensity of bubble collapse increase, leading to enhanced
594 physical effects such as higher velocity and stronger shear
595 forces in the bulk liquid^{17, 31}. On the other hand, sonication
596 time only varies the duration of the induce shear force.
597 Higher intensity ultrasound can also lower the threshold of
598 droplet break-up, as determined by the Weber number:

$$599 \quad N_{We} = \rho_c d \overline{v^2} / \tau \quad (12)$$

601

Where $\overline{v^2}$ is the average of the square of the relative velocity between the emulsion droplets and the surrounding continuous phase across the flow field, d is the droplet size, ρ_c is the density of the continuous phase, and σ is the interfacial tension between the water-oil interface. As the intensity of shear stress increases, the Weber number increases, and droplet deformation is promoted³⁶. In comparison, prolonged sonication time has no effect on the intensity of shear stress.

Interestingly, as the droplet size starts reaching the steady state transition region, the impact of the power input on energy efficiency appears to lessen (shown as triangular data points in Fig. 7). In this case the data points from different power inputs can be fitted to a single power-law correlation. In addition, the impact of sonication time reduces, reflected by the power-law index of around -0.6. These results indicate that the relative impact of sonication time and power input gradually converges, giving a new correlation between the droplet size and energy density:

$$d[4,3] = E_v^{-0.6} \quad (13)$$

The power law index of -0.6 corresponds well to results from previous studies^{8, 24}. It is also consistent with the theory proposed by Hinze and Kolmogorov (equation (2)) that predicts a power-law index of around -0.4 for power density ϵ at the late steady state transition region, where the droplet size almost reaches the breakup-recoalescence balance^{21, 22, 29, 30}. Although steady state values were not obtained here, it can be predicted from the equation (2) that as the increase of the power density, the droplets size in the steady state will decrease, which also confirmed in previous work, where the size decrease from 0.5 μm at 62 W to 0.3 μm at 225 W²⁴.

Fig. 7 Volume mean diameter $d[4,3]$ data presented in Fig. 2, replotted as a function of energy density E_v . Triangular symbols are used for the data points in the steady state transition region. The power-law indices of the two regions are -1.1 and -0.6 before and during the steady state transition region, respectively. The inset plot presents the correlation between $d[4,3]/E_v$ and calorimetric power P_{cal} .

Comparison of the emulsion droplets size reduction kinetics and energy efficiency between US and HPH

A comparison of the kinetics of EDS reduction was conducted between ultrasound and high-pressure homogenisation, where a similar approach was used to investigate the impact of the relevant HPH parameters. The pressure drop across the valve determined by the pressure gauge, ΔP , and the number of processing pass (N) were used to quantify the effective energy delivered to the fluid¹⁴. The correlation between droplet size and ΔP was shown in Fig. 8A. Across all passes the power-law indices was approximately -0.8. Similarly, power law indices of approximately -0.8 were obtained across all pressures for correlations between droplet size and pass number (Fig. 8B). In contrast to the US system, all the results from the HPH system could be fitted using a similar power law index. There was no significant deviation across the tested range of pressure and passes. The power-law correlation was also fitted to the data from Gupta et al., which spanned up to 20 passes²¹. A power-law index around -0.7 was obtained from $N=2$ to $N=6$. Beyond 10 passes significant deviations occurred at high ΔP , indicating the approach to a shear stress limit. For the data obtained beyond 10 passes a power law index of approximately -0.4 was obtained.

Fig. 8 Volume mean diameter, $d[4,3]$ after high pressure homogenisation. The same data are presented as a function of ΔP (A) and number of processing pass, N (B). Error bars represent the standard deviation of triplicate experiments.

To compare the energy efficiency of the two emulsification systems the droplet size data are plotted together as a function of energy density (Fig. 9A and 9B). The energy density was calculated on the basis of i) the effective energy delivered to the emulsion, $E_{delivered}$ (Fig. 9A) and ii) the total

electricity consumed by the emulsification device, $E_{consumed}$ (Fig. 9B). The former indicates the effectiveness of droplet size reduction while the latter provides the operating efficiency of the emulsification equipment used in this study. The delivered energy density ($E_{delivered}$) was determined based on calorimetric power (P_{calo}) for the US system, and pressure gauge readings (ΔP) for the HPH system. The consumed energy density ($E_{consumed}$) was determined using a power meter that provided direct measurements of the overall electric power drawn by the units during operation.

Fig. 9 $d[4,3]$ data from Fig.s 2 and 8 replotted as a function of delivered energy density $E_{delivered}$ (A, power density based on P_{calo} and ΔP) and consumed energy density $E_{consumed}$ (B, power density calculated based on total electric power). The triangular symbols with the solid trendline represent the droplet size data in the steady state transition region of the US batch emulsification. (C) Size distribution of droplets obtained by ultrasonication (50% amplitude for 20 min and continuous

flow system at 3.8 mL min^{-1}) and high-pressure homogenisation ($\Delta P = 101 \text{ MPa}$ at 3 and 4 passes)

Compared to both US systems, relatively small droplets were obtained using HPH across the tested range of pressure and passes. As demonstrated previously, ΔP and N have equivalent impact meaning the data points could be fitted to a single power-law correlation with an index of -0.9 (the difference between this value and the -0.8 presented previously is due to the inclusion of the additional of energy used in a UT pre-emulsification step (20 J mL^{-1})). This value is between the power-law indices for US prior to (-1.1) and during the transition region (-0.6). For an $E_{delivered} \sim 190 \text{ J mL}^{-1}$, the two systems produced emulsions with a similar $d[4,3]$ of around 300 nm . However, with an increase in $E_{delivered}$, the US system went into the steady state transition region, in which the efficiency on droplet size reduction decreases. In comparison, the HPH system is able to sustain the same efficiency throughout the tested range, reaching a droplet size of $\sim 120 \text{ nm}$ at 425 J mL^{-1} compared to the US system requiring around 790 J mL^{-1} to reach the same size. The difference is mainly due to differences in the maximum shear stress in two systems as discussed previously. For US continuous flow system, a larger droplets size with an increased power-law index of -0.65 was observed. Compared to the US batch system, the larger droplets size can be attributed to the insufficient residence time within the energy dissipation zone (red-lined area in Fig. 1B). Although the power density, ϵ , increased compared to batch system, from 0.66 W mL^{-1} to 166.67 W mL^{-1} (processing volume decreased from 60 mL to 0.3 mL), the residence time at highest energy density was 9 s , which was considerably short compared with the batch system. Additionally, the increase of the residence time in CF system with a higher power density results in an increased power-law index (less efficiency) compared to the batch system, indicating the change in the geometry of the processing unit, which may vary the shape of the cavitation zone and pattern of turbulence, could potentially alter the efficiency of emulsification.

For all the systems, the consumed energy density (Fig. 9B) was greater than the delivered energy density (Fig. 9A) due to inherent inefficiencies associated with the delivery of energy to the system and the background and operating power requirements. As previously observed for the US system, higher power inputs were more energy efficient, and this trend was even more pronounced on the basis of $E_{consumed}$. For the HPH, whereas the energy efficiency is independent of ΔP based on $E_{delivered}$, the energy efficiency in terms of $E_{consumed}$ was considerably improved at higher pressures/fewer passes due to the addition of operation power from the high-pressure pump and background power. These results demonstrate that by increasing operating pressure, higher energy effectiveness can be achieved compared to increasing the number of processing passes. For example, a similar volume-mean diameter around 400 nm can be obtained by processing for one pass

at 86 MPa with an $E_{consumed}$ of 945 J mL⁻¹, or using 4 passes at 24 MPa, with a much higher $E_{consumed}$ of 2575 J mL⁻¹. Similar droplet size distributions were obtained using US batch system (50% amplitude at 20 min) and HPH (101 MPa for 3 or 4 passes) (Fig. 9C). The volume-mean diameters of the three distributions are 125 nm (US), 128 nm (HPH 101 MPa for 3 passes) and 120 nm (HPH 101 MPa for 4 passes). A slightly narrower size distribution was obtained using HPH than US. This could be due to the shear stress differences between the two systems, especially as the US system has entered the steady state transition region. US CF system, however, shows a relatively wider size distribution with a shoulder at size range 5 μ m at the similar energy density range compared to US batch system ($E_{delivered}$ ~700 J mL⁻¹ and $E_{consumed}$ ~1500 J mL⁻¹), indicating an insufficient homogenization process. Kentish et al. also stated that the effective design is of great importance for US CF system to achieve a higher emulsification efficiency, as they attributed the reason of the shoulder on the size distribution was due the fluid bypassing of the cavitation zone with a similar US CF setup employed³³. Table 1 compares the ratio of power delivered for droplets reduction (P_{calo} for US and ΔP for HPH) to the power

consumed on emulsification unit measured from the power meter reading (the error stands for the reading fluctuations during the operation). The power input on processing unit stands for the power consumed on the generation of ultrasonic waves, $P_{sonication}$, and pressure increase of HPH, $P_{increase}$, which was determined by subtracting the background power reading (i.e. when the sonication unit is on but not delivering ultrasound, and when the HPH is operating at 0 MPa) from the power consumption during operation. The ratio is similar for both units (0.63-0.67 for US, and 0.59-0.64 for HPH). Much of the energy inefficiency for the HPH unit was due to background power, which is exaggerated for a small lab-scale unit. Yap et al. have compared the energy consumption of laboratory-scale (GEA Niro Soavi NS1001), pilot-scale (GEA Niro Soavi NS3030) and industrial-scale (GEA Niro Soavi NS5355) HPHs³⁷. It was found that there was a 3-fold decrease in energy consumption upon scaling-up from lab- to industrial-scale. When this scale up factor is applied to the results obtained with the current HPH system, the energy efficiency of HPH at 86 MPa is similar to the lab-scale US system at 1.47 W/mL (Fig. 10).

Table 1 Power delivery comparison between emulsification unit consumption and power delivered for droplets reduction in US and HPH systems.

US			HPH		
P_{calo}/W	$P_{sonication}/W$	$P_{calo}/P_{sonication}$	$\Delta P/W$	$P_{increase}/W$	$\Delta P/P_{increase}$
8.5 $\pm$ 1.1	13.6 $\pm$ 0.4	0.63	55.2 $\pm$ 3.2	93.8 $\pm$ 4.3	0.59
15.6 $\pm$ 0.9	24.4 $\pm$ 0.5	0.64	110.8 $\pm$ 5.4	172.4 $\pm$ 4.6	0.64
23.3 $\pm$ 1.2	35.3 $\pm$ 0.7	0.66	155.3 $\pm$ 7.6	242.3 $\pm$ 6.5	0.64
31.4 $\pm$ 1.0	47.2 $\pm$ 1.1	0.67	207.6 $\pm$ 10.2	322.6 $\pm$ 11.6	0.64
41.5 $\pm$ 1.2	64.1 $\pm$ 0.9	0.65	265.3 $\pm$ 11.2	412.1 $\pm$ 10.7	0.64

716

Fig. 10 $d[4,3]$ as a function of $E_{consumed}$ for a lab-scale US and an industrial-scale HPH system. The energy density of the HPH system was derived from the data presented in Fig. 9 using a scale-up energy efficiency conversion factor as discussed in the text. The US data is the most energy efficient result as presented in Fig. 9B.

For US systems, although a number of studies of large-scale reactors have been done for various applications³⁸⁻⁴⁰ that show the potential of operating at larger scales, a comparison of the energy efficiency across different scales of operation has

yet to be performed. This would be of great interest to explore in future studies, especially in emulsion systems.

Conclusions

In this work, experiments were conducted to better understand the kinetics of ultrasonic emulsification in relation to operational parameters and energy efficiency. Empirical correlations were developed from the data that could be practically applied for the prediction of EDS. The impact of sonication time, power input and processing volume were investigated separately. Two kinetic regions of size reduction (the fast size reduction region and the steady state transition region) were distinguished based on the power-law fit. Within the fast size reduction region (F-region), the power input had a greater impact on size reduction than sonication time. Combining the individual parameters, the emulsion droplets size was correlated to energy density on the basis of calorimetric power. Power-law indices of -1.1 and -0.6 were obtained for the initial and transition regions, respectively. The difference in the impact of power input and sonication time was minimal in the transition region. The power-law index of -1.1

represents a faster size reduction compared to previous studies, indicating better emulsification capability of US than previously reported. A comparison of emulsification effectiveness and energy efficiency was made between US and HPH systems. The HPH system was able to obtain smaller droplet sizes than US, indicating a slightly higher effectiveness of size reduction. At lab-scale the US system could produce nanoemulsions using less electrical energy than HPH, indicating better energy efficiency at this scale. Taking the up-scaling factor for HPH into consideration, the tested US system still shows a competitive emulsification efficiency. Due to the unique combination of physical/chemical effects offered by ultrasound, the application of ultrasonic emulsification can be usefully applied to other fields including emulsion polymerisation and nutrient encapsulation where not only the physical shear, but also the radical formation from the cavitation is important. However, more attention needs to be put on the design of continuous ultrasonic flow systems that have good capabilities at large-scale.

Conflicts of interest

There are no conflicts to declare.

Acknowledgements

This research was supported under the University of Melbourne for the award of MRS, Australian Research Council's Industrial Transformation Research Program (ITRP) funding scheme (project number IH120100005). The ARC Dairy Innovation Hub is a collaboration between The University of Melbourne, The University of Queensland and Dairy Innovation Australia Ltd.

References

1. T. F. Tadros, *Applied surfactants: principles and applications*, John Wiley & Sons, 2006.
2. B. P. Binks, *Modern aspects of emulsion science*, Royal Society of Chemistry, 1998.
3. T. F. Tadros, *Journal*, 1983.
4. H. Karbstein and H. Schubert, *Chemical Engineering and Processing: Process Intensification*, 1995, **34**, 205-211.
5. S. Abbas, K. Hayat, E. Karangwa, M. Bashari and X. Zhang, *Food Engineering Reviews*, 2013, **5**, 139-157.
6. D. J. McClements and J. Rao, *Critical reviews in food science and nutrition*, 2011, **51**, 285-330.
7. A. Shanmugam and M. Ashokkumar, *Food Hydrocolloids*, 2014, **39**, 151-162.
8. T. Leong, T. Wooster, S. Kentish and M. Ashokkumar, *Ultrasonics Sonochemistry*, 2009, **16**, 721-727.
9. A. Forgiarini, J. Esquena, C. Gonzalez and C. Solans, *Langmuir*, 2001, **17**, 2076-2083.
10. P. Izquierdo, J. Esquena, T. F. Tadros, C. Dederen, M. Garcia, N. Azemar and C. Solans, *Langmuir*, 2002, **18**, 26-30.
11. I. Sole, A. Maestro, C. González, C. Solans and J. M. Gutiérrez, *Langmuir*, 2006, **22**, 8326-8332.
12. I. N. Seekkumarachchi, K. Tanaka and H. Kumazawa, *Industrial & engineering chemistry research*, 2006, **45**, 372-390.
13. A. Håkansson, L. Fuchs, F. Innings, J. Revstedt, C. Trägårdh and B. Bergenstahl, *Chemical Engineering Communications*, 2013, **200**, 93-114.
14. S. Schultz, G. Wagner, K. Urban and J. Ulrich, *Chemical Engineering & Technology*, 2004, **27**, 361-368.
15. S. M. Jafari, Y. He and B. Bhandari, *Journal of Food Engineering*, 2007, **82**, 478-488.
16. Y. Y. J. Zuo, P. Hébraud, Y. Hemar and M. Ashokkumar, *Ultrasonics Sonochemistry*, 2012, **19**, 421-426.
17. J. Canselier, H. Delmas, A. Wilhelm and B. Abismail, *Journal of Dispersion Science and Technology*, 2002, **23**, 333-349.
18. M. Ashokkumar, R. Bhaskaracharya, S. Kentish, J. Lee, M. Palmer and B. Zisu, *Dairy Science & Technology*, 2010, **90**, 147-168.
19. T. Leong, M. Ashokkumar and S. Kentish, *Acoust. Aust*, 2011, **39**, 54-63.
20. M. Ashokkumar, *Theoretical and experimental sonochemistry involving inorganic systems*, Springer Science & Business Media, 2010.
21. A. Gupta, V. Narsimhan, T. A. Hatton and P. S. Doyle, *Langmuir*, 2016, **32**, 11551-11559.
22. A. Gupta, H. B. Eral, T. A. Hatton and P. S. Doyle, *Soft matter*, 2016, **12**, 1452-1458.
23. M. Li and H. Fogler, *Journal of Fluid Mechanics*, 1978, **88**, 499-511.
24. B. Abismai, J. Canselier, A. Wilhelm, H. Delmas and C. Gourdon, *Ultrasonics Sonochemistry*, 2000, **7**, 187-192.
25. J. Hinze, *AIChE Journal*, 1955, **1**, 289-295.
26. A. Kolmogorov, 1949.
27. S. Tcholakova, N. Vankova, N. D. Denkov and T. Danner, *Journal of colloid and interface science*, 2007, **310**, 570-589.
28. N. Vankova, S. Tcholakova, N. D. Denkov, V. D. Vulchev and T. Danner, *Journal of colloid and interface science*, 2007, **313**, 612-629.
29. N. Vankova, S. Tcholakova, N. D. Denkov, I. B. Ivanov, V. D. Vulchev and T. Danner, *Journal of Colloid and Interface Science*, 2007, **312**, 363-380.
30. P. Walstra, *Chemical Engineering Science*, 1993, **48**, 333-349.
31. O. Behrend and H. Schubert, *Ultrasonics Sonochemistry*, 2001, **8**, 271-276.
32. S. G. Gaikwad and A. B. Pandit, *Ultrasonics Sonochemistry*, 2008, **15**, 554-563.
33. S. Kentish, T. Wooster, M. Ashokkumar, S. Balachandran, R. Mawson and L. Simons, *Innovative Food Science & Emerging Technologies*, 2008, **9**, 170-175.
34. A. Cucheval and R. Chow, *Ultrasonics Sonochemistry*, 2008, **15**, 916-920.

- 859 35. L. E. Kinsler, A. R. Frey, A. B. Coppens and J. V.
860 Sanders, *Fundamentals of Acoustics, 4th Edition*, by
861 Lawrence E. Kinsler, Austin R. Frey, Alan B. Coppens,
862 James V. Sanders, pp. 560. ISBN 0-471-84789-5.
863 Wiley-VCH, December 1999., 1999, 560.
- 864 36. B. Tal-Figiel, *Chemical Engineering Research and*
865 *Design*, 2007, **85**, 730-734.
- 866 37. B. H. Yap, G. J. Dumsday, P. J. Scales and G. J.
867 Martin, *Bioresource technology*, 2015, **184**, 280-
868 285.
- 869 38. Y. Asakura, T. Nishida, T. Matsuoka and S. Koda,
870 *Ultrasonics Sonochemistry*, 2008, **15**, 244-250.
- 871 39. P. R. Gogate, V. S. Sutkar and A. B. Pandit, *Chemical*
872 *Engineering Journal*, 2011, **166**, 1066-1082.
- 873 40. B. Zisu, R. Bhaskaracharya, S. Kentish and M.
874 Ashokkumar, *Ultrasonics Sonochemistry*, 2010, **17**,
875 1075-1081.

876