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1 *Title page*

2 **Effects of 3,4-dimethylpyrazole phosphate (DMPP) on nitrification and the abundance**  
3 **and community composition of soil ammonia oxidizers in three land uses**

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10 **Abstract**

11 The application of the nitrification inhibitor, 3,4-dimethylpyrazole-phosphate (DMPP) is  
12 considered as an effective strategy to mitigate agricultural nitrogen loss. However, the  
13 inhibitory effect of DMPP on nitrification is variable and the importance of the soil microbial  
14 community to the variability is poorly understood. In this study, nine soils were collected  
15 across three land uses to investigate the impact of DMPP on nitrification and associated  
16 dynamics of ammonia oxidizers in a 28-day microcosm incubation. The results showed that  
17 the efficacy of DMPP at inhibiting net nitrification rates varied highly from no effect to 63.6%  
18 during the first week of incubation. The abundance of ammonia-oxidizing bacteria (AOB),  
19 rather than ammonia-oxidizing archaea (AOA) was significantly correlated with nitrate  
20 concentrations across three land uses and significantly inhibited by DMPP addition. DMPP  
21 had higher efficacy in neutral and alkaline wheat and vegetable soils, compared with pasture  
22 soils. Canonical correspondence analysis suggested that soil pH was the most influential  
23 factor explaining the community composition of AOB and AOA in the collected soils.  
24 However, neither ammonium nitrate nor DMPP addition had a significant effect on the  
25 community composition of AOB or AOA during the incubation indicated by non-metric  
26 multidimensional scaling ordination. Taken together, our findings indicated that DMPP  
27 slowed nitrification by inhibiting the growth of AOB, and DMPP application affected the  
28 abundance of AOB more than the ammonia oxidizers community composition.

29 **Keywords:**

30 3,4-dimethylpyrazole-phosphate; ammonia-oxidizing archaea; ammonia-oxidizing bacteria;  
31 abundance; community composition

32

33

## 34 **Introduction**

35 Nitrogen (N)-based fertilizer is applied extensively in agriculture as an essential element for  
36 plant growth (Gilsanz et al. 2016). According to the prediction of the international fertilizer  
37 industry association (IFA), the global demand for N fertilizers will continuously increase and  
38 reach approximate 161 million metric tonnes in 2018 (IFA, 2014). However, a major concern  
39 associated with the use of synthetic N fertilizers is the substantial N losses into the  
40 environment and the low nitrogen use efficiency (NUE) (Chen et al. 2008). Furthermore, the  
41 rising costs of synthetic N fertilizers have strengthened farmers' commercial incentive to  
42 reduce N fertilization and as a consequence N losses (Ernfors et al. 2014). Nitrification is a  
43 critical component in the N cycle, which causes N losses in the form of nitrate ( $\text{NO}_3^-$ )  
44 leaching and nitrous oxide ( $\text{N}_2\text{O}$ ) emissions from agricultural systems (Wakelin et al. 2014;  
45 Chen et al. 2015). Nitrate leaching could trigger waterway eutrophication and drinking water  
46 contamination (Philippot et al. 2007).  $\text{N}_2\text{O}$  is a potent greenhouse gas, with about 300-fold  
47 greater warming potential than  $\text{CO}_2$  on a per molecule basis, and is also involved in  
48 destruction of the stratospheric ozone layer (Hu et al. 2015a). Therefore, major efforts have  
49 been devoted to decreasing nitrification rates to retain N in the soil for plant uptake (Ernfors  
50 et al. 2014; Wakelin et al. 2014).

51 The incorporation of nitrification inhibitors (NIs) into N fertilizers is widely adopted as an  
52 efficiency-enhanced strategy to mitigate N losses through reducing nitrification rates  
53 (Akiyama et al. 2009). NIs are chemical compounds capable of modifying the active site of  
54 ammonia monooxygenase (AMO) and causing the formation of polypeptides, thus resulting  
55 in the inactivation of AMO activity and delay in nitrification (McCarty and Bremner 1989).  
56 Ammonia oxidation, as the rate-limiting step of nitrification, is catalyzed by AMO encoded  
57 by the *amoA* gene within two distinct groups of nitrifying microbes: ammonia-oxidizing  
58 bacteria (AOB) and ammonia-oxidizing archaea (AOA) (Brochier-Armanet et al. 2008;

59 Purkhold et al. 2000). Of the available NIs, 3,4-dimethylpyrazole-phosphate (DMPP) has  
60 been widely used together with synthetic N fertilizers (Weiske et al. 2001; Di and Cameron  
61 2011; Liu et al. 2015) and animal manure (Hatch et al. 2005; Fangueiro et al. 2009; Maienza  
62 et al. 2014). To achieve comparable effects, DMPP requires much lower application rates  
63 than other NIs such as dicyandiamide (DCD) (Chaves et al. 2005; Chen et al. 2008). More  
64 importantly, no negative effect of DMPP on non-target microorganisms in soils has been  
65 reported (Tindaon et al. 2012; Kong et al. 2016). The inhibition of nitrification by DMPP was  
66 previously thought to be through indiscriminately binding to the complex of membrane-  
67 bound proteins and inactivating the AMO-containing AOB (Kleineidam et al. 2011;  
68 Benckiser et al. 2013). This perception, however, was challenged by the recent studies of  
69 Dong et al. (2013) and Florio et al. (2014), which provided evidence that DMPP inhibited  
70 AOA as well. Therefore, the underlying contribution of AOA and AOB to the DMPP-  
71 induced inhibition of nitrification is still in debate.

72 DMPP has been frequently applied in pasture and farming systems, but the inhibitory  
73 efficacy was reported to differ largely within and across agricultural systems. For example,  
74 studies found that the effect of DMPP on nitrification varied from no effect to 28.2% in  
75 grasslands (Fangueiro et al. 2009; Di and Cameron 2012), from 8.6 to 56.6% in cropping  
76 systems (Weiske et al. 2001; Liu et al. 2013; Migliorati et al. 2014), and from 13.8 to 19.1%  
77 in vegetable production systems (Xu et al. 2005). Environmental variables such as  
78 temperature and moisture have been widely studied and considered as key factors controlling  
79 the efficacy of DMPP at inhibiting nitrification (Azam et al. 2001; Menéndez et al. 2009;  
80 2012 ). Recently, Shi et al. (2016) suggested that the efficacy of NIs at reducing the  
81 nitrification rates was highly correlated with the abundance and activity of AOB. In contrast,  
82 McGeough et al. (2016) found that the efficacy of NIs was largely affected by soil properties  
83 such as soil clay and organic matter content. To date, there is limited knowledge regarding

84 the importance of soil microbial community composition to the variable efficacy of DMPP at  
85 inhibiting nitrification, which has greatly hindered our ability to assess the efficacy of DMPP  
86 in different scenarios.

87 The main objective of this study was to explore the influence of ammonia oxidizers and  
88 land uses on DMPP efficacy. Nine soils across three land uses (pasture, wheat and vegetable)  
89 were incubated following fertilizer application in presence and absence of DMPP in a 28-day  
90 microcosm incubation at 25°C. The following hypotheses were tested: (1) The abundance  
91 and/or community composition of AOA and AOB would respond differently to DMPP  
92 application, owing to their different physiological characteristics and enzyme systems for  
93 ammonia oxidation; (2) The efficacy of DMPP at inhibiting nitrification would be higher in  
94 the wheat and vegetable soils compared to the pasture soils, due to their lower organic matter  
95 content.

## 96 **Materials and methods**

### 97 **Site descriptions**

98 Soils samples were collected in October, 2015 from nine sites across three land uses (pasture,  
99 wheat, and vegetable) in Victoria and New South Wales, Australia. Pasture soils were  
100 collected from Dookie (P-DK, 36°25' S, 145°42' E), Glenormiston (P-GN, 38°15' S, 142°52'  
101 E) and Finley (P-FL, 35°38' S, 145°40' E). Two wheat growing soils were collected from  
102 Finley (W-FL<sub>1</sub>, 35°33' S, 145°32' E; W-FL<sub>2</sub>, 35°39' S, 145°37' E) and one from Horsham  
103 (W-HS, 36°45' S, 142°07' E). Vegetable growing soils were collected at Clyde (V-CD,  
104 38°08' S, 145°20' E), Werribee (V-WB, 37°56' S, 144°40' E) and Perry Bridge (V-PB,  
105 37°59' S, 147°16' E). The soils were classified using the Australian Soil Classification (Isbell  
106 2002).

107 **Soil sampling and physicochemical characterization**

108 At each site, four surface soil samples (0-10 cm) were mixed together as a composite sample  
109 and passed through a 2 mm sieve. Gravimetric water content was determined by oven-drying  
110 sub-samples at 105°C for 24 h. Soil pH was determined using a ratio of 1:5 (fresh soil : water)  
111 with an Orion Star A211 pH-meter (Thermo Scientific Inc., Melbourne, Australia). Soil  
112 organic matter was determined using the  $K_2Cr_2O_7$  wet oxidation and colorimetric method as  
113 previously described by Walkley and Black (1934). Soil exchangeable  $NH_4^+$ -N and  $NO_3^-$ -N  
114 were extracted using a ratio of 1:5 (fresh soil : 1 M KCl, w/v) by shaking at 180 rpm for 1 h,  
115 and the solution, filtered through a Whatman 42 filter paper, was measured by a Segmented  
116 Flow Analyzer (SAN++, Skalar, Breda, Holland). Total N was determined using the classic  
117 Dumas method of combustion on an isotope ratio mass spectrometer (Sercon Hydra, Crewe,  
118 UK). Particle size analysis was measured using the sieve and hydrometer procedures. Details  
119 of the soil physical and chemical properties are shown in Table 1.

120 **Soil microcosm incubation**

121 The microcosm incubation was established to assess the effects of DMPP addition on  
122 nitrification and nitrifiers in the nine collected soils. Three treatments involving (1) control,  
123 (2)  $NH_4NO_3$ , and (3)  $NH_4NO_3$  plus DMPP were set up in 250 ml plastic vials containing 20 g  
124 of soils (oven dry-weight equivalent) with three replicates. Nitrogen was added at a rate of 75  
125 mg  $NH_4^+$ -N  $kg^{-1}$  soil and 75 mg  $NO_3^-$ -N  $kg^{-1}$  soil. DMPP was applied in the form of solution  
126 at 1% of the applied  $NH_4^+$ -N according to the typical recommended rate (Serna et al. 2000;  
127 Pasda et al. 2001). All microcosms were pre-incubated for a week at 25°C and then adjusted  
128 to 60% water-filled pore space (WFPS) after adding treatment solutions (or deionized water  
129 for the control treatment). The final bulk density varied between 1.07 and 1.28  $g\ cm^{-3}$ . The  
130 vials were closed immediately after treatment solutions were applied and incubated at 25°C  
131 in the dark for 28 days. During the incubation, the microcosms were aerated through opening

132 the vials for about three min every three days and moisture loss was replenished every week.  
133 Soil samples (10 g) were collected from three vials on days 0, 1, 4, 7, 14 and 28 for  
134 determination of exchangeable  $\text{NH}_4^+$ -N and  $\text{NO}_3^-$ -N. Soils (0.5 g) were collected from the  
135 same vials on days 0, 7 and 28 for DNA extraction and molecular analysis.

136 The equation developed by Persson and Wirén (1995) was used to calculate the net  
137 nitrification rates in the first week and for the incubation time (28 days):

$$138 \quad n_{(d0-d7)} = [(\text{NO}_3^- \text{-N})_{d7} - (\text{NO}_3^- \text{-N})_{d0}] / 7, \quad (1)$$

$$139 \quad n_{(d0-d28)} = [(\text{NO}_3^- \text{-N})_{d28} - (\text{NO}_3^- \text{-N})_{d0}] / 28, \quad (2)$$

140 where  $(\text{NO}_3^- \text{-N})_{d0}$ ,  $(\text{NO}_3^- \text{-N})_{d7}$  and  $(\text{NO}_3^- \text{-N})_{d28}$  are the  $\text{NO}_3^-$ -N concentrations in the soil on  
141 days 0, 7 and 28, respectively.

#### 142 **DNA extraction and quantitative PCR (qPCR) analysis**

143 Total genomic DNA was isolated from 0.25 g of soil using MoBio PowerSoil™ DNA  
144 Isolation Kits (MoBio Laboratories, Carlsbad, CA, USA) according to the manufacturer's  
145 instructions. The concentration and quality of the extracted DNA were assessed using the  
146 NanoDrop ND2000c Spectrophotometer (NanoDrop Technologies, Wilmington, DE, USA).  
147 The abundance of AOB and AOA *amoA* genes were quantified on a Bio-Rad CFX96 Optical  
148 Real-Time PCR Detection System (Bio-Rad, Laboratories Inc, Hercules, CA, USA) using  
149 the primer sets *amoA1F/amoA2R* (Rotthauwe et al. 1997) and *CrenamoA23f/CrenamoA616r*  
150 (Tourna et al. 2008), respectively. The 10  $\mu\text{l}$  reaction mixture contained 5  $\mu\text{l}$  of SYBR Premix  
151 Ex Taq™ (TaKaRa Biotechnology, Otsu, Shiga, Japan), 0.2  $\mu\text{l}$  of each primer (20  $\mu\text{M}$ ), and  
152 1-10 ng of template DNA. Thermal cycling conditions for the AOB *amoA* gene were as  
153 follows: 30 s at 95°C, 40 cycles of 30 s at 95°C, 30 s at 56°C, and 30 s at 72°C. As for the  
154 AOA *amoA* gene, thermal cycling conditions were: 30 s at 95°C, 40 cycles of 30 s at 95°C,  
155 45 s at 55°C, and 45 s at 72°C. Standard curves were generated using ten-fold serial dilutions  
156 of plasmids containing correct inserts of the targeted genes. Melting curve analysis was

157 performed between 75 and 94.5°C at the end of each amplification assay to evaluate the  
158 specificity of qPCR products. The amplification efficiencies for all qPCR runs ranged  
159 between 95-105%.

160 **Community profiling of the *amoA* gene by terminal restriction fragment length**  
161 **polymorphism (T-RFLP) analysis**

162 The community composition of AOB and AOA in soils on days 0 and 7 of the incubation  
163 were determined by T-RFLP analysis of the *amoA* genes, using the fluorescently labelled  
164 primers FAM-CrenamoA23f/CrenamoA616r and FAM-amoA1F/amoA2R, respectively (Hu  
165 et al. 2015b). The 25 µl PCR reaction mixture contained 2 µl of diluted template DNA (1-10  
166 ng), 0.3 µl of each primer (20 µM), 2.5 µl of 10 × NH<sub>4</sub> reaction buffer, 1 µl of MgCl<sub>2</sub> solution  
167 (50 mM), 1 µl of BSA (20 mM), 0.5 µl of dNTP mix (20 mM), and 1.5 units of BioTaq DNA  
168 polymerase (Bioline, Sydney, Australia). The thermal cycling conditions were as follows:  
169 95°C for 5 mins; 35 cycles of 30 s at 95°C, 45 s at 55°C, 45 s at 72°C, followed by 10 min at  
170 72°C. The PCR products were purified using the Wizard SV Gel and PCR Clean-Up System  
171 (Promega, San Luis Obispo, CA, USA) and quantified using the NanoDrop ND2000c  
172 Spectrophotometer.

173 The restriction digestion was carried out in a 10 µl mixture containing 300 ng of purified  
174 PCR products, 0.1 µl of BSA, 1 µl of 10 × NEBuffer, and 5 units of the restriction enzymes  
175 MspI for AOB or RsaI (BioLabs, Sydney, Australia) for AOA. Digests were incubated at  
176 37°C for 3 h, and then denatured for 10 min at 95°C. Terminal restriction fragments (TRFs)  
177 were size-separated with an ABI PRISM 3500 Genetic analyzer (Applied Biosystems, CA,  
178 USA), and analyzed using a local southern size calling method (peaks > 50 bp) and a peak  
179 amplitude threshold setting of 50, using Genemapper version 4.0 (Applied Biosystems).  
180 TRFs with peak height comprising less than 1% of the total peak height were removed from  
181 the downstream analysis, and peaks that differed by less than 1 bp were combined into the

182 same TRF (Singh and Thomas 2006; Hu et al. 2015b). The relative fluorescence abundances  
183 of all TRFs were exported for the analysis of community composition.

#### 184 **Statistical analysis**

185 The *amoA* gene copy numbers were log-transformed prior to statistical analysis to meet  
186 normality assumptions. One-way analysis of variance (ANOVA) was used to compare net  
187 nitrification rates and the *amoA* gene abundance across treatments in SPSS 19.0 (IBM, USA).  
188 Spearman's correlation was performed to assess the correlations between the *amoA* gene  
189 copy numbers and NO<sub>3</sub><sup>-</sup>-N concentrations. Difference at  $P < 0.05$  was considered to be  
190 statistically significant. Canonical correspondence analysis was performed with the Monte  
191 Carlo permutation test (the number of permutations is 499) to investigate the significant  
192 environmental factors influencing the community composition of AOB and AOA based on  
193 the T-RFLP data from soils collected on day 0 using CANOCO 4.5 (Microcomputer Power,  
194 Ithaca, NY, USA). Non-metric multidimensional scaling (NMDS) was used to visualize the  
195 Bray-Curtis dissimilarity matrices based on the T-RFLP data from soils collected on day 7 of  
196 the incubation. Permutational multivariate analysis of variance (PERMANOVA) was  
197 performed to examine the significance of Bray-Curtis dissimilarity between treatments, by  
198 using the *adonis* function of the *vegan* package in R platform.

#### 199 **Results**

##### 200 **Soil physical and chemical properties**

201 The physical and chemical properties of the soil samples were highly variable across the nine  
202 sampling sites (Table 1). Briefly, soil pH values varied from 5.44 in V-PB to 7.96 in V-WB,  
203 and mineral N contents shifted between 10.6 and 31.8 µg g<sup>-1</sup> soil for exchangeable NH<sub>4</sub><sup>+</sup>-N  
204 and 8.3 and 78.5 µg g<sup>-1</sup> soil for NO<sub>3</sub><sup>-</sup>-N. Soil organic matter ranged widely from 1.35 to 7.30%  
205 and total N ranged from 0.05 to 0.49%, with higher values recorded in P-FL and P-DK soils.

206 The wide range of the shift in soil properties provided a good opportunity to explore the  
207 effect of DMPP across various soil orders and the three land uses.

### 208 **Dynamics of mineral N and net nitrification rates**

209 In the control treatments, the exchangeable  $\text{NH}_4^+$ -N concentrations remained low and almost  
210 unchanged around a level of  $10 \mu\text{g g}^{-1}$  soil for all the examined soils (Fig. 1). In the  $\text{NH}_4\text{NO}_3$   
211 treatments, the exchangeable  $\text{NH}_4^+$ -N concentrations decreased over time, with dramatic  
212 declines towards the control levels in P-DK, P-FL, W-FL<sub>1</sub>, W-HS, V-CD and V-WB soils  
213 within two weeks, while DMPP amendment significantly slowed down the decline of  
214 exchangeable  $\text{NH}_4^+$ -N in these soils ( $P < 0.05$ ). However, regardless of DMPP addition, the  
215 exchangeable  $\text{NH}_4^+$ -N concentrations in P-GN, W-FL<sub>2</sub>, and V-PB soils remained much  
216 higher than those in the control treatments until the end of the incubation at day 28.

217 The  $\text{NO}_3^-$ -N concentrations gradually increased in the control microcosms (Fig. 2),  
218 indicative of the oxidation of  $\text{NH}_3$  derived from the mineralization of native organic N.  
219 Addition of  $\text{NH}_4\text{NO}_3$  significantly increased the  $\text{NO}_3^-$ -N concentrations over time ( $P < 0.05$ ),  
220 which coincided with the dramatic declines in exchangeable  $\text{NH}_4^+$ -N, indicating the  
221 occurrence of nitrification. In treatments amended with DMPP, nitrification was evidently  
222 reduced by DMPP at certain sampling points. The net nitrification rates for all soils attained  
223 the highest values in the  $\text{NH}_4\text{NO}_3$  treatments in the first week of incubation, and DMPP  
224 exerted significant inhibition on the net nitrification rates in P-FL, W-FL<sub>1</sub>, W-HS, V-CD and  
225 V-WB soils ( $P < 0.05$ ), which varied between 23.4 and 63.6% (Table 2). The net nitrification  
226 rates substantially decreased at the end of incubation, and no significant difference was  
227 observed in treatments with and without DMPP.

### 228 **Changes in the abundance of AOB and AOA**

229 The abundance of AOB and AOA in the V-PB soil was below the detection limit of qPCR  
230 analysis, therefore is not shown in Fig. 3. The abundance of AOB in the control treatment

231 varied greatly from  $2.62 \times 10^4$  copies  $g^{-1}$  soil in W-FL<sub>2</sub> soil to  $8.13 \times 10^8$  copies  $g^{-1}$  soil in V-  
232 WB soil, and was evidently higher in vegetable soils than in pasture and wheat soils, except  
233 for P-FL soil (Fig. 3). Addition of  $NH_4NO_3$  significantly increased AOB abundance in W-HS  
234 and V-WB soils on day 7, compared with the control treatments, while DMPP significantly  
235 decreased the AOB abundance by 69.1% and 51.4%, respectively. In the  $NH_4NO_3$  treatments,  
236 the abundance of AOB was observed to significantly increase in P-GN, P-FL, W-HS, V-CD  
237 and V-WB soils on day 28 ( $P < 0.05$ ). However, significant decreases in the AOB abundance  
238 due to DMPP addition were only found in W-HS and V-WB soils on day 28 ( $P < 0.05$ ). The  
239 AOA abundance was highest in the vegetable soils ranging from  $1.55 \times 10^7$  to  $1.35 \times 10^8$   
240 copies  $g^{-1}$  soil, and much lower in the pasture and wheat soils ranging between  $5.73 \times 10^4$  and  
241  $2.18 \times 10^6$  copies  $g^{-1}$  soil (Fig. 4). However, there was no significant difference in the AOA  
242 abundance between treatments during the incubation.

243 Pearson correlation analysis revealed that AOB abundances showed significant correlations  
244 with  $NO_3^-$ -N concentrations in pasture soils ( $R^2 = 0.44$ ,  $P < 0.001$ ), wheat soils ( $R^2 = 0.32$ ,  $P$   
245  $< 0.01$ ) and vegetable soils ( $R^2 = 0.31$ ,  $P < 0.05$ ). However, significant correlation between  
246 AOA abundance and  $NO_3^-$ -N was only observed in pasture soils ( $R^2 = 0.30$ ,  $P < 0.01$ ) (Fig. 5).

#### 247 **Effects of DMPP on the community composition of AOB and AOA**

248 The community composition of AOB and AOA in soils collected on days 0 and 7 of the  
249 incubation was analyzed using T-RFLP. After being digested by the MspI enzyme, the AOB  
250 *amoA* gene yielded 14 distinct TRFs across land uses. Canonical correspondence analysis  
251 (CCA) showed that soil organic matter ( $F = 2.24$ ,  $P = 0.002$ ) and soil pH ( $F = 2.02$ ,  $P = 0.002$ )  
252 were the two most influential soil properties explaining the AOB community composition,  
253 with the first two axes of the CCA accounting for 63.9 % of the total variance. The CCA  
254 biplots showed that the AOB community from the same land use formed the same cluster  
255 (Fig. 6). NMDS ordinations based on the Bray-Curtis dissimilarity matrices revealed a clear

256 divergence of AOB assemblages between different soils (Fig. 7), which was further  
257 corroborated by the PERMANOVA analysis (Pseudo- $F = 6.7$ ,  $P < 0.001$ ).

258 The AOA *amoA* gene digested by the *RsaI* enzyme yielded nine TRFs. Canonical  
259 correspondence analysis showed that soil pH ( $F = 1.49$ ,  $P = 0.014$ ) and indigenous  $\text{NO}_3^-$ -N ( $F$   
260  $= 1.46$ ,  $P = 0.040$ ) were the significant environmental factors explaining the AOA  
261 community composition, with the first two axes standing for 52.9 % of the total AOA  
262 variance (Fig. 6). NMDS ordinations showed no clear divergence between treatments (Fig. 7).  
263 The PERMANOVA analysis suggested that the land use had a significant impact on the AOA  
264 community profiling composition (Pseudo- $F = 2.4$ ,  $P < 0.001$ ).

## 265 **Discussion**

### 266 **DMPP inhibits nitrification through impairing the abundance of AOB**

267 We found that the DMPP played a significant inhibitory role in the soils with pH ranging  
268 between 6.28 and 7.96, whereas no DMPP-related inhibition on nitrification was observed in  
269 soils with pH between 5.44 and 5.65 (Tables 1 and 2). Recent studies suggested that soil pH is  
270 a critical factor driving the niche partitioning of AOB and AOA (as reviewed in Hu et al.  
271 2014). On one hand, neutral and slightly alkaline soils were generally considered as  
272 favourable habitats for AOB growth (Shen et al. 2008; Jia and Conrad 2009), and AOB could  
273 not be capable of oxidizing  $\text{NH}_3$  under extremely low soil pH conditions (De Boer and  
274 Kowalchuk 2001). The explanation for the inability of AOB to nitrify  $\text{NH}_3$  under acid  
275 conditions is the exponential ionization of  $\text{NH}_3$  to  $\text{NH}_4^+$  in acid soils (Hu et al. 2014). In this  
276 study, the highest AOB abundance was observed in the vegetable soils V-CD (pH 7.02) and  
277 V-WB (pH 7.96) soils, whereas the lowest AOB abundance appeared in P-GN (pH 5.55) and  
278 W-FL2 (pH 5.49) soils (Fig. 3). On the other hand, numerous studies indicated that AOA  
279 plays a more dominant role in mediating nitrification in acid soils, with comprehensive  
280 evidence provided by the  $^{13}\text{CO}_2$ -DNA-SIP and 454 barcoded pyrosequencing techniques

281 (Zhang et al. 2012; Hu et al. 2013). In this study, a higher abundance of AOA than AOB was  
282 only found in W-FL<sub>2</sub> soil with the lowest pH of 5.49 among the investigated soils. Although a  
283 higher abundance of AOB than AOA was observed in other acid soils including P-GN, W-  
284 FL<sub>2</sub> and V-PB soils, previous researches revealed that the numerical advantage at genomic  
285 level does not necessarily equal the dominance at functional level (Jia and Conrad 2009). It  
286 should be noted that the nitrification rate in P-DK soil was much higher than other acid soils,  
287 and could be probably ascribed to the oxidation of NH<sub>3</sub> derived from the mineralization of  
288 native organic N, as suggested by the rapid increase in nitrate content in the control treatment.  
289 This is probably facilitated by the higher organic N in P-DK soil (Table 2). However, no  
290 significantly inhibitory effect of DMPP was observed in these acid soils during the incubation.  
291 Taken together, it could be inferred that DMPP has an inhibitory effect on soil nitrification  
292 through impairing the AOB abundance favoured by the higher soil pH conditions, whereas  
293 DMPP was not efficient at decreasing nitrification in acid soils.

294 The different responses of AOB and AOA to DMPP inhibition could be ascribed to not only  
295 the pH-associated niche differentiation, but the different enzyme systems for nitrification.  
296 Firstly, it is conventionally considered that hydroxylamine (NH<sub>2</sub>OH) acts as the intermediate  
297 of ammonia oxidation (NH<sub>3</sub> → NH<sub>2</sub>OH), catalyzed by the AMO enzyme, and DMPP could  
298 inhibit nitrification through inactivation of the AMO activity (Hu et al. 2015a). However, the  
299 homologous gene encoding the hydroxylamine oxidoreductase (HAO), which is a necessity  
300 for the oxidation of NH<sub>2</sub>OH, has not been identified in AOA genomes (Tournia et al. 2011). It  
301 is speculated that AOA could possess a unique way of nitrification, unlike AOB. One  
302 alternative is that AOA could produce nitroxyl hydride (HNO) as an intermediate of  
303 ammonia oxidation (NH<sub>3</sub> → HNO) (Walker et al. 2010). Secondly, the annotated AOA *amo*  
304 sequences were more similar to the genes encoding bacterial particulate methane  
305 monooxygenases (pMMO) rather than AOB *amoA* genes, indicative of functional differences

306 in AMO between the AOA and AOB, which is responsible for the oxidation of  $\text{NH}_3$  to  
307  $\text{NH}_2\text{OH}$  (Walker et al. 2010). An early study reported that DMPP reduced AMO activity,  
308 while pMMO stayed unimpaired in an agricultural crop rotation (Weiske et al. 2001). Thirdly,  
309 the possibility of heterotrophic growth has been suggested for AOA (Tourna et al. 2011),  
310 such as group I.1c, well-known as a major lineage of AOA (Lehtovirta et al. 2009). The  
311 potential of heterotrophic nitrification was further suggested by Hu et al. (2013) who found  
312 that the group I.1c abundance had a significant positive correlation with soil organic C.  
313 Therefore, DMPP could inhibit nitrification by inhibiting AOB growth, and the inefficiency  
314 of DMPP on AOA-mediated nitrification could be caused by the different enzyme systems  
315 and the mixotrophic growth of AOA.

#### 316 **Environmental and soil factors influencing DMPP efficacy**

317 It has been more than a decade since the earliest report regarding the use of DMPP to regulate  
318 N supply through maintaining N as  $\text{NH}_4^+$  and reducing  $\text{NO}_3^-$  leaching, consequently  
319 improving NUE (Serna et al. 2000; Zerulla et al. 2001), decreasing greenhouse gas emissions  
320 and increasing crop yields (Pasda et al. 2001; Weiske et al. 2001). Given these benefits  
321 derived from DMPP application, the factors influencing the efficacy of DMPP have attracted  
322 considerable research interest. Soil temperature has been considered as one of the most  
323 important factors controlling DMPP efficacy, because temperature has the great potential to  
324 impact the mobility of DMPP and the degradation rate of the DMPP in soils (Azam et al.  
325 2001; Irigoyen et al. 2003). In our study, DMPP slowed down the decline of exchangeable  
326  $\text{NH}_4^+$ -N in one or two weeks, and the longest extension was observed in W-FL<sub>1</sub> and W-HS  
327 soils till the end of incubation at 25°C (Figs. 1 and 2), which was comparable with previous  
328 studies that showed DMPP retained exchangeable  $\text{NH}_4^+$ -N for eight days at 30°C, and 40  
329 days at 20°C (Irigoyen et al. 2003; Menéndez et al. 2012). However, regardless of DMPP  
330 addition, the exchangeable  $\text{NH}_4^+$ -N concentration remained at high levels in P-GN, W-FL<sub>2</sub>

331 and V-PB soils till day 28. The different changes in exchangeable  $\text{NH}_4^+$ -N under the same  
332 temperature indicate that there are other critical factors influencing the efficacy of DMPP in  
333 the retention of  $\text{NH}_4^+$ -N.

334 The effect of soil properties on DMPP efficacy was first reported by Barth et al. (2001)  
335 through a short-term incubation experiment, suggesting that soil textural properties,  
336 specifically the proportion of clay, are important factors influencing the binding behaviour of  
337 positively charged DMPP. In this study, the inhibitory effects of DMPP on net nitrification  
338 rates were generally higher in the soils with clay fractions less than 10% (W-HS, V-CD and  
339 P-FL soils). This is well supported by the previous findings that the inhibitory effects of  
340 DMPP were lower in soils with high proportions of clay (Wu et al. 2007; Barth et al. 2008),  
341 probably because more DMPP is adsorbed by soil components, resulting in less amount of  
342 DMPP to inhibit nitrification (Roco and Blu 2006). However, DMPP inhibited nitrification  
343 by 40.9% in V-WB soil in the first week, even with 36% clay content in this study (Tables 1  
344 and 2). As suggested by McGeough et al. (2016), clay sorption capacity was not only  
345 dependent on the proportion of clay fraction, but also the type of clay minerals. Therefore, the  
346 proportion and type of clay in soils should be determined to better understand their effects on  
347 DMPP efficiencies in future studies. Apart from clay fractions, soil organic matter is  
348 suggested as another important soil component involved in the sorption of organic  
349 compounds (Kelliher et al. 2008; McGeough et al. 2016). In this study, the soil organic  
350 matter content is generally higher in the pasture soils than in the wheat and vegetable soils  
351 (Table 1), consequently, the higher inhibitory effect of DMPP on net nitrification found in the  
352 latter two soils could be due to the lower organic matter (Table 2). Recently, Marsden et al.  
353 (2016) reported that the sorption of DMPP was higher in the soil with higher amount of  
354 organic matter with the  $^{14}\text{C}$  labelled DMPP applied in three distinct soils. The sorption  
355 phenomenon by soil organic matter also applies to other NIs, as a meta-analysis conducted by

356 Kelliher et al. (2008) showed that the amount of DCD sorption onto soil organic matter soils  
357 is a critical factor determining DCD efficacy. A recent study showed that the efficiency of  
358 nitrapyrin, diminished over the long-term addition of soil organic matter (Fisk et al. 2015).

### 359 **Conclusions**

360 In conclusion, through analysis of functional gene abundance and community composition  
361 analysis in microcosm incubation of nine soils across three land uses, we found that only  
362 AOB was significantly decreased by DMPP addition, accompanied by the decreased net  
363 nitrification rates. The effectiveness of DMPP at reducing nitrification was higher in neutral  
364 and alkaline soils wheat and vegetable soils than in pasture soils, indicating DMPP could play  
365 an inhibitory role in nitrification through impairing AOB growth. We provided evidence that  
366 the community composition of ammonia oxidizers were more resilient than gene abundance  
367 in response to DMPP addition. Our results indicated that soil clay fractions and organic  
368 matter content could be critical factors influencing DMPP efficacy. Further studies are  
369 needed to determine the changes in metabolic activities of AOB and AOA after DMPP  
370 addition to provide more direct evidence.

### 371 **Acknowledgements**

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550 **Table 1** The physical and chemical properties for the soil samples collected from nine sites in three land uses in Australia

Soil properties	P-DK	P-GN	P-FL	W-FL <sub>1</sub>	W-FL <sub>2</sub>	W-HS	V-CD	V-WB	V-PB
Land use	Pasture	Pasture	Pasture	Wheat	Wheat	Wheat	Vegetable	Vegetable	Vegetable
Location	Dookie	Glenormiston	Finley	Finley	Finley	Horsham	Clyde	Werribee	Perry Bridge
Mean annual temperature (°C)	20.9	19.1	22.9	22.9	22.9	22.7	19.9	20.5	20.3
Mean annual rainfall (mm)	550	779	430	430	430	414	812	538	628
Soil Order	Chromosol	Dermosol	Sodosol	Sodosol	Sodosol	Vertosol	Podosol	Chromosol	Podosol
pH	5.65	5.55	6.28	6.37	5.49	6.37	7.02	7.96	5.44
Soil organic matter (%)	5.96	5.27	7.30	5.33	2.61	3.67	3.11	1.73	1.35
Exchangeable NH <sub>4</sub> <sup>+</sup> -N (µg g <sup>-1</sup> soil)	17.7	12.1	21.0	31.8	15.8	12.0	12.4	14.0	10.6
NO <sub>3</sub> <sup>-</sup> -N (µg g <sup>-1</sup> soil)	37.8	9.3	36.9	78.5	16.7	8.3	32.3	27.9	9.0
Total N (%)	0.40	0.21	0.49	0.31	0.19	0.20	0.30	0.17	0.05
Particle size (%)									
Clay (< 0.002 mm)	11	19	8	11	40	3	4	36	5
Silt (0.002-0.02 mm)	30	14	15	44	11	37	11	23	5

Sand (0.02-2 mm)

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551 **Table 2** The average net nitrification rates of the three treatments across the three land uses  
 552 during the first week of incubation and the 28-day of incubation

Soils	Time intervals	Net nitrification rates			% Inhibition by DMPP
		Control	NH <sub>4</sub> NO <sub>3</sub>	NH <sub>4</sub> NO <sub>3</sub> + DMPP	
P-DK	d <sub>0</sub> -d <sub>7</sub>	3.22 ± 0.14 a	11.16 ± 0.52 b	11.61 ± 0.21 b	nil
	d <sub>0</sub> -d <sub>28</sub>	3.15 ± 0.12 a	5.26 ± 0.15 b	5.10 ± 0.07 b	nil
P-GN	d <sub>0</sub> -d <sub>7</sub>	0.78 ± 0.26 a	2.58 ± 0.14 b	2.71 ± 0.35 b	nil
	d <sub>0</sub> -d <sub>28</sub>	0.87 ± 0.09 a	1.63 ± 0.27 b	1.39 ± 0.02 b	nil
P-FL	d <sub>0</sub> -d <sub>7</sub>	2.88 ± 0.14 a	11.90 ± 0.15 c	9.12 ± 0.34 b	23.4
	d <sub>0</sub> -d <sub>28</sub>	2.28 ± 0.01 a	4.50 ± 0.02 b	4.18 ± 0.24 b	nil
W-FL <sub>1</sub>	d <sub>0</sub> -d <sub>7</sub>	2.37 ± 0.62 a	4.37 ± 0.30 b	2.42 ± 0.99 a	44.6
	d <sub>0</sub> -d <sub>28</sub>	1.66 ± 0.32 a	3.14 ± 0.14 b	2.97 ± 0.34 b	nil
W-FL <sub>2</sub>	d <sub>0</sub> -d <sub>7</sub>	1.88 ± 0.17 a	3.64 ± 0.08 b	2.88 ± 0.40 b	20.9
	d <sub>0</sub> -d <sub>28</sub>	1.27 ± 0.05 a	2.41 ± 0.05 b	2.42 ± 0.06 b	nil
W-HS	d <sub>0</sub> -d <sub>7</sub>	1.62 ± 0.12 a	9.42 ± 0.16 b	3.43 ± 0.95 a	63.6
	d <sub>0</sub> -d <sub>28</sub>	1.62 ± 0.06 a	3.56 ± 0.16 b	3.37 ± 0.25 b	nil
V-CD	d <sub>0</sub> -d <sub>7</sub>	2.08 ± 0.22 a	11.70 ± 0.26 c	8.76 ± 0.67 b	25.1
	d <sub>0</sub> -d <sub>28</sub>	2.22 ± 0.07 a	4.12 ± 0.08 b	4.00 ± 0.08 b	nil
V-WB	d <sub>0</sub> -d <sub>7</sub>	5.11 ± 0.47 a	14.06 ± 0.87 c	8.31 ± 0.62 b	40.9
	d <sub>0</sub> -d <sub>28</sub>	2.67 ± 0.15 a	4.54 ± 0.21 b	4.13 ± 0.17 b	nil
V-PB	d <sub>0</sub> -d <sub>7</sub>	0.30 ± 0.12 a	0.63 ± 0.20 a	0.63 ± 0.11 a	nil
	d <sub>0</sub> -d <sub>28</sub>	0.42 ± 0.02 a	0.56 ± 0.19 a	0.57 ± 0.03 a	nil

553 The rates were expressed as mean ± standard errors. The unit is µg g<sup>-1</sup> soil day<sup>-1</sup>. Different  
 554 letters in the same row denote significant difference at the *P* < 0.05 level; nil denotes no  
 555 observed inhibition derived from DMPP addition.

556 **Figure legends**

557 **Fig. 1** Changes in the exchangeable  $\text{NH}_4^+$ -N concentrations of the nine soils in the three land  
558 uses (pasture, wheat and vegetable) during the 28-day microcosm incubation. Error bars  
559 represent standard errors of three replicates.

560 **Fig. 2** Changes in the  $\text{NO}_3^-$ -N concentrations of the nine soils in the three land uses (pasture,  
561 wheat and vegetable) during the soil microcosm incubation. Error bars represent standard  
562 errors of three replicates.

563 **Fig. 3** Changes in the AOB abundance during the 28-day microcosm incubation of soils in  
564 the three land uses (pasture, wheat and vegetable) during the 28-day microcosm incubation.  
565 Error bars represent standard errors of three replicates. Note: Y axes scales differ between  
566 charts.

567 **Fig. 4** Changes in the AOA abundance during the 28-day microcosm incubation of soils in  
568 the three land uses (pasture, wheat and vegetable) during the 28-day microcosm incubation.  
569 Error bars represent standard errors of three replicates. Note: Y axes scales differ between  
570 charts.

571 **Fig. 5** Relationship between AOB or AOA abundance and the  $\text{NO}_3^-$ -N concentrations during  
572 the 28-day microcosm incubation.

573 **Fig. 6** CCA ordination biplots indicating the relationship between the community  
574 composition AOB or AOA and soil properties in the three land uses (pasture, wheat and  
575 vegetable) before incubation. Arrows indicate the direction and magnitude of measurable  
576 variables associated with the microbial community composition.

577 **Fig. 7** Nonmetric multi-dimensional scaling ordinations based on the Bray-Curtis  
578 dissimilarity matrices of the T-RFLP data of AOB digested using the MspI enzyme and AOA

579 digested using the RsaI enzyme across different treatments (C, control; F,  $\text{NH}_4\text{NO}_3$ ; FD,  
580  $\text{NH}_4\text{NO}_3$  + DMPP) and land uses (P, pasture; W, wheat; V, vegetable) on day 7 of the  
581 incubation.